

Lecture 1 Semiconductor Physics “Review”

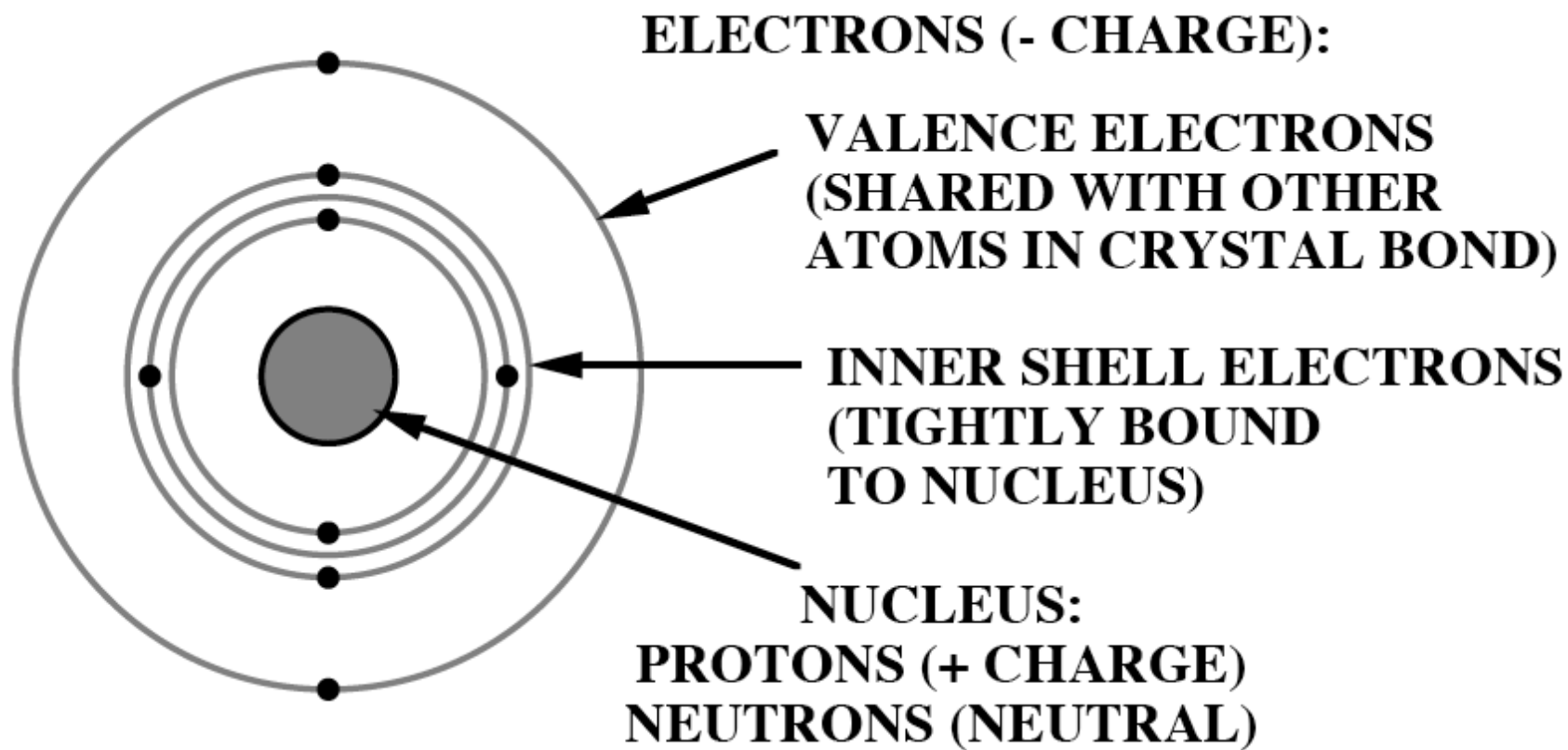
- **P-N Junction Diode**
 - **Reverse bias diode junction capacitance**
- **Text: Johns & Martin Sec. 1.1 pp. 1-12**

Semiconductor Physics “Review”

- **Chemistry Review**
- **Types of Materials**
 - **Conductor, Insulator, Semiconductor**
- **PN junction**
 - **Depletion region**
 - **Junction capacitance**

Chemistry Review

- **Model of atom (ridiculously oversimplified)**



Conduction

- **Conductor**
 - Each atom: valence e- not used in bond
 - “Sea of electrons”
 - Small applied $V \Rightarrow$ Lots of charge moving $\Rightarrow$ Large $I \Rightarrow$ Low resistance
- **Insulator**
 - All valence e- tightly bound
 - Applied $V \Rightarrow$ Little charge moving $\Rightarrow$ Small $I \Rightarrow$ High resistance
- **Semiconductor**
 - “In between”

Periodic Table

METALS														NONMETALS								
PERIODS	I A		TRANSITION METALS														VII A	0				
	1	2	III B	IV B	VB	VIB	VII B	VIII		IB	II B	III A	IV A	V A	VI A	1	2					
1	1.0079 H [1]																	1.0079 H [1]	4.00260 He [2]			
2	6.941 Li [3]	9.01218 Be [4]													10.81 B [5]	12.011 C [6]	14.0067 N [7]	15.9994 O [8]	18.9984 F [9]	20.179 Ne [10]		
3	22.9898 Na [11]	24.305 Mg [12]													26.9815 Al [13]	28.086 Si [14]	30.9738 P [15]	32.06 S [16]	35.453 Cl [17]	39.948 Ar [18]		
4	39.098 K [19]	40.08 Ca [20]	44.9559 Sc [21]	47.90 Ti [22]	50.9414 V [23]	51.996 Cr [24]	54.9380 Mn [25]	55.847 Fe [26]	58.9332 Co [27]	58.71 Ni [28]	63.546 Cu [29]	65.38 Zn [30]	69.72 Ga [31]	72.59 Ge [32]	74.9216 As [33]	78.96 Se [34]	79.904 Br [35]	83.80 Kr [36]				
5	85.4678 Rb [37]	87.62 Sr [38]	88.9059 Y [39]	91.22 Zr [40]	92.9064 Nb [41]	95.94 Mo [42]	98.9062 Tc [43]	101.07 Ru [44]	102.9055 Rh [45]	106.4 Pd [46]	107.868 Ag [47]	112.40 Cd [48]	114.82 In [49]	118.69 Sn [50]	121.75 Sb [51]	127.60 Te [52]	126.9045 I [53]	131.30 Xe [54]				
6	132.9054 Cs [55]	137.34 Ba [56]	[57-71] *	178.49 Hf [72]	180.9479 Ta [73]	183.85 W [74]	186.2 Re [75]	190.2 Os [76]	192.22 Ir [77]	195.09 Pt [78]	198.9065 Au [79]	200.59 Hg [80]	204.37 Tl [81]	207.2 Pb [82]	208.9804 Bi [83]	(210) Po [84]	(210) At [85]	(222) Rn [86]				
7	(223) Fr [87]	(226.0254) Ra [88]	[89-103] †	[104]	[105]	[106]	[107]	[108]														

* LANTHANIDE
SERIES

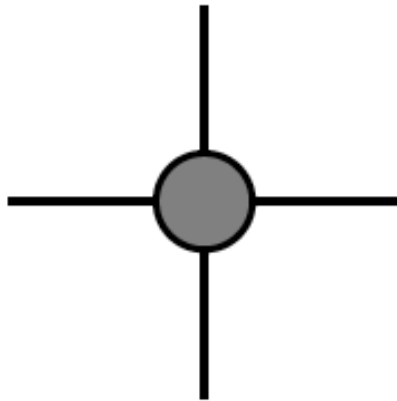
† ACTINIDE
SERIES

138.9055 La [57]	140.12 Ce [58]	140.9077 Pr [59]	144.24 Nd [60]	(145) Pm [61]	150.4 Sm [62]	151.96 Eu [63]	157.25 Gd [64]	158.9254 Tb [65]	162.50 Dy [66]	164.9304 Ho [67]	167.26 Er [68]	168.9342 Tm [69]	173.04 Yb [70]	174.97 Lu [71]
18, 9, 2	20, 8, 2	21, 8, 2	22, 8, 2	23, 8, 2	24, 8, 2	25, 8, 2	25, 9, 2	27, 8, 2	28, 8, 2	29, 8, 2	30, 8, 2	31, 8, 2	32, 8, 2	32, 9, 2
(227) Ac [89]	232.0381 Th [90]	231.0359 Pa [91]	238.0289 U [92]	237.0482 Np [93]	(242) Pu [94]	(243) Am [95]	(245) Cm [96]	(245) Bk [97]	(248) Cf [98]	(253) Es [99]	(254) Fm [100]	(256) Md [101]	(253) No [102]	(257) Lr [103]
18, 9, 2	18, 10, 2	20, 9, 2	21, 9, 2	23, 9, 2	24, 8, 2	25, 8, 2	25, 9, 2	26, 9, 2	28, 8, 2	29, 8, 2	30, 8, 2	31, 8, 2	32, 8, 2	32, 9, 2

Source: General Chemistry, ISBN 0-669-63362-3

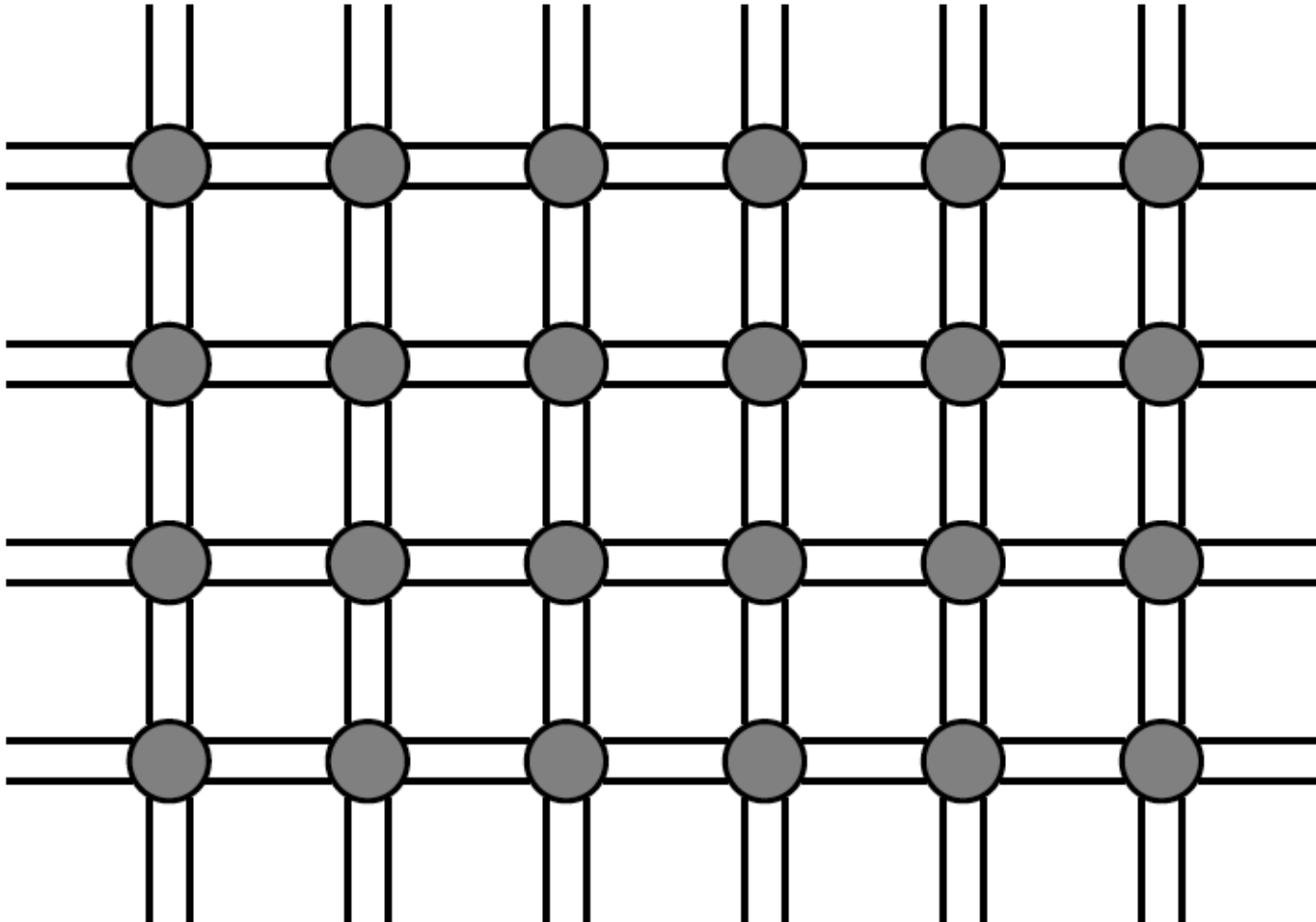
Bonding model

- **Lines represent valence electrons**
- **Silicon**
4 valence electrons



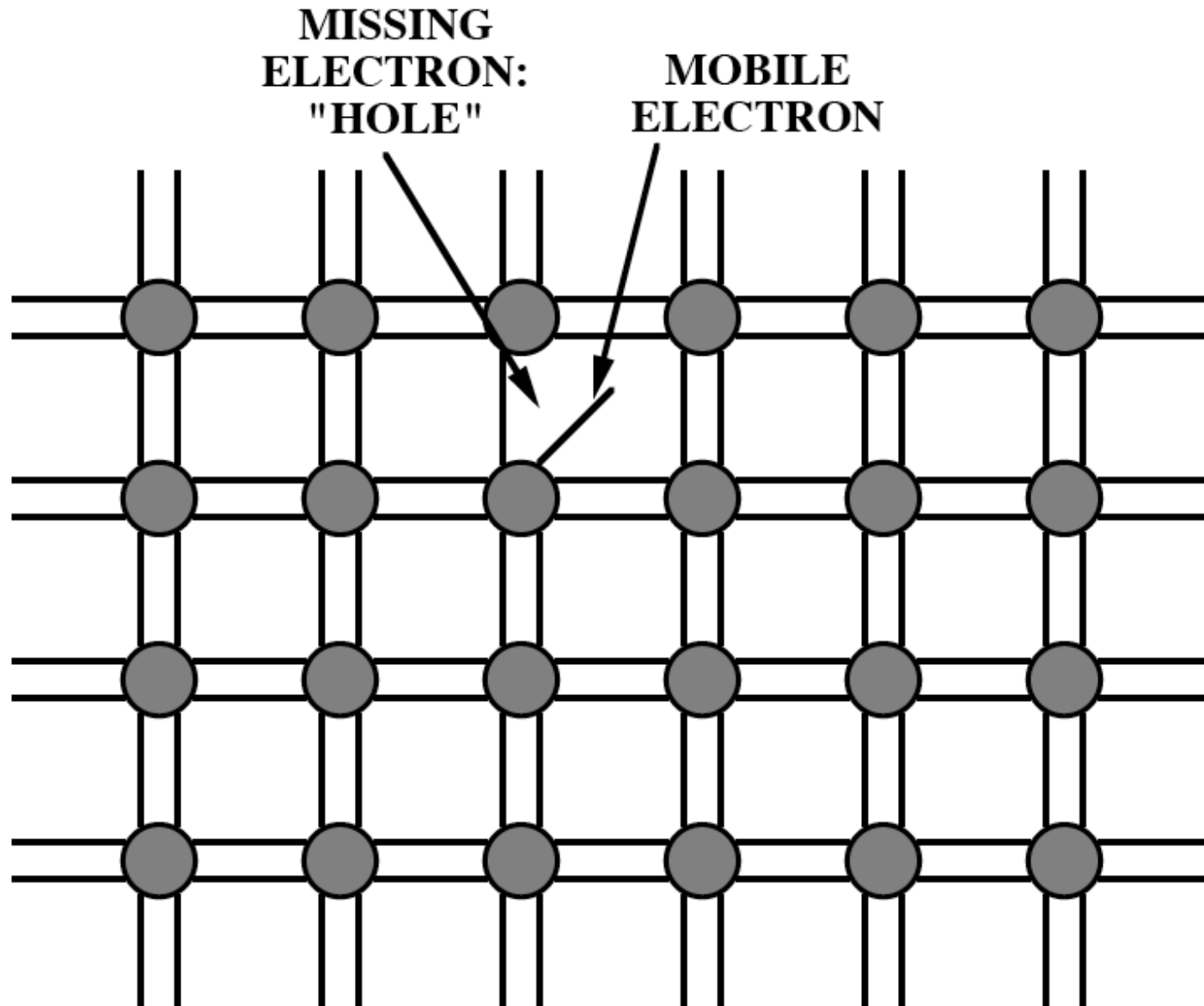
Pure (Intrinsic) Silicon, $T = 0 \text{ }^{\circ}\text{K}$ (Absolute Zero)

All valence electrons tightly bound



Pure Silicon, $T = 300\text{ }^{\circ}\text{K}$ (Room Temperature)

Thermal energy frees some valence electrons



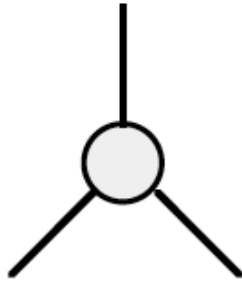
Pure Silicon, $T = 300\text{ }^{\circ}\text{K}$ (Room Temperature)

- **Thermal energy frees some valence electrons**
 - **“electron”**: mobile negative charge
 - **Missing electrons (“holes”)** behaves as mobile positive charge
- **Equal number of holes, electrons**
- **Relatively poor conductor**

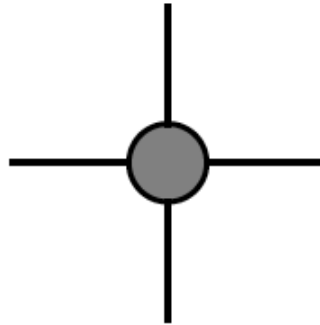
Doping

- Intentionally introduce impurity atoms to unbalance number of holes, electrons
- Adjacent columns in periodic table

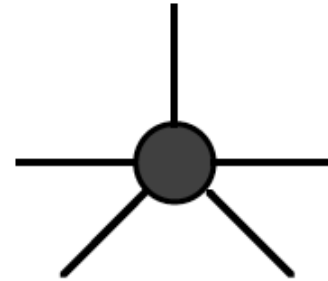
Boron



Silicon

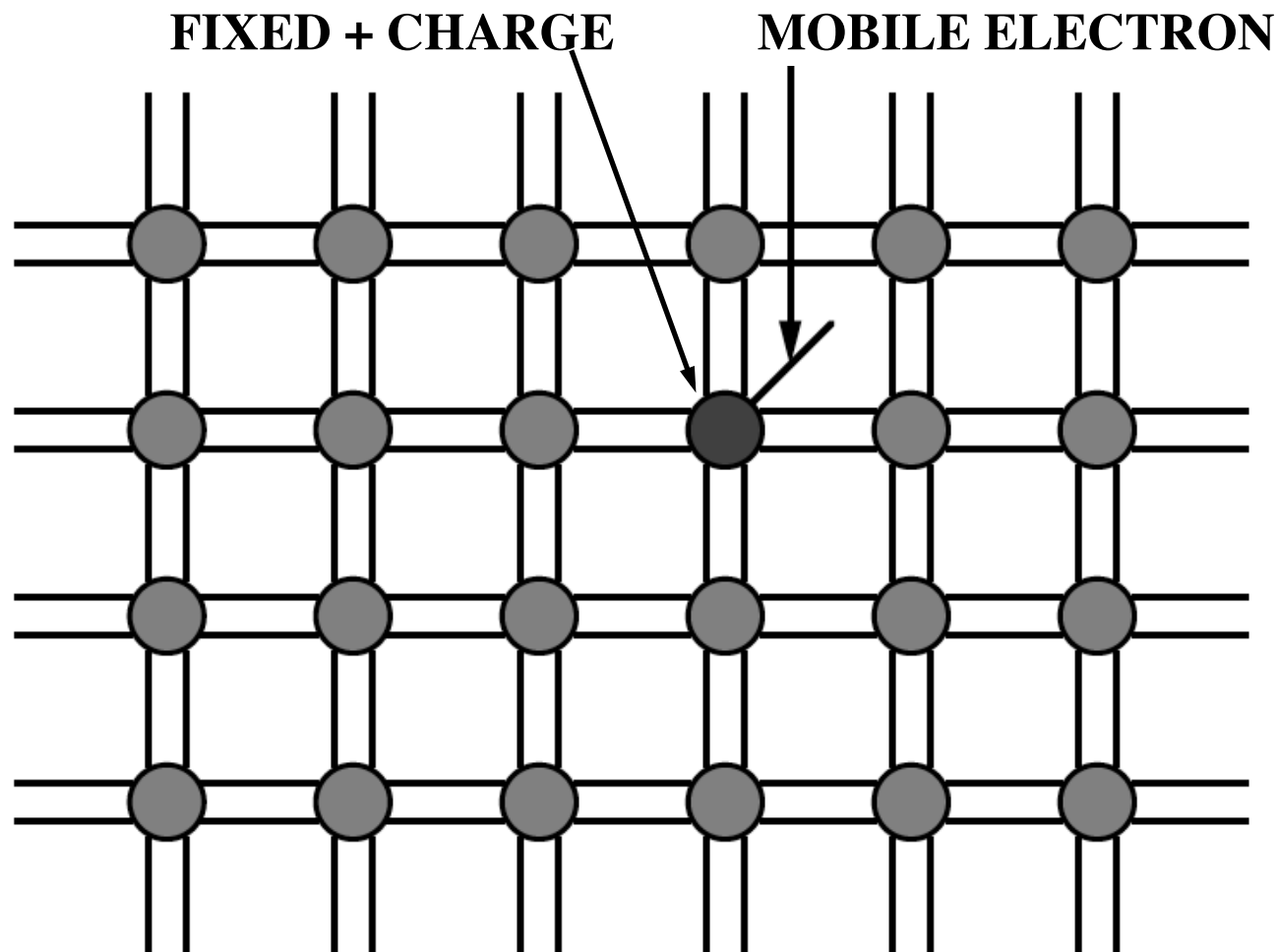


Phosphorous



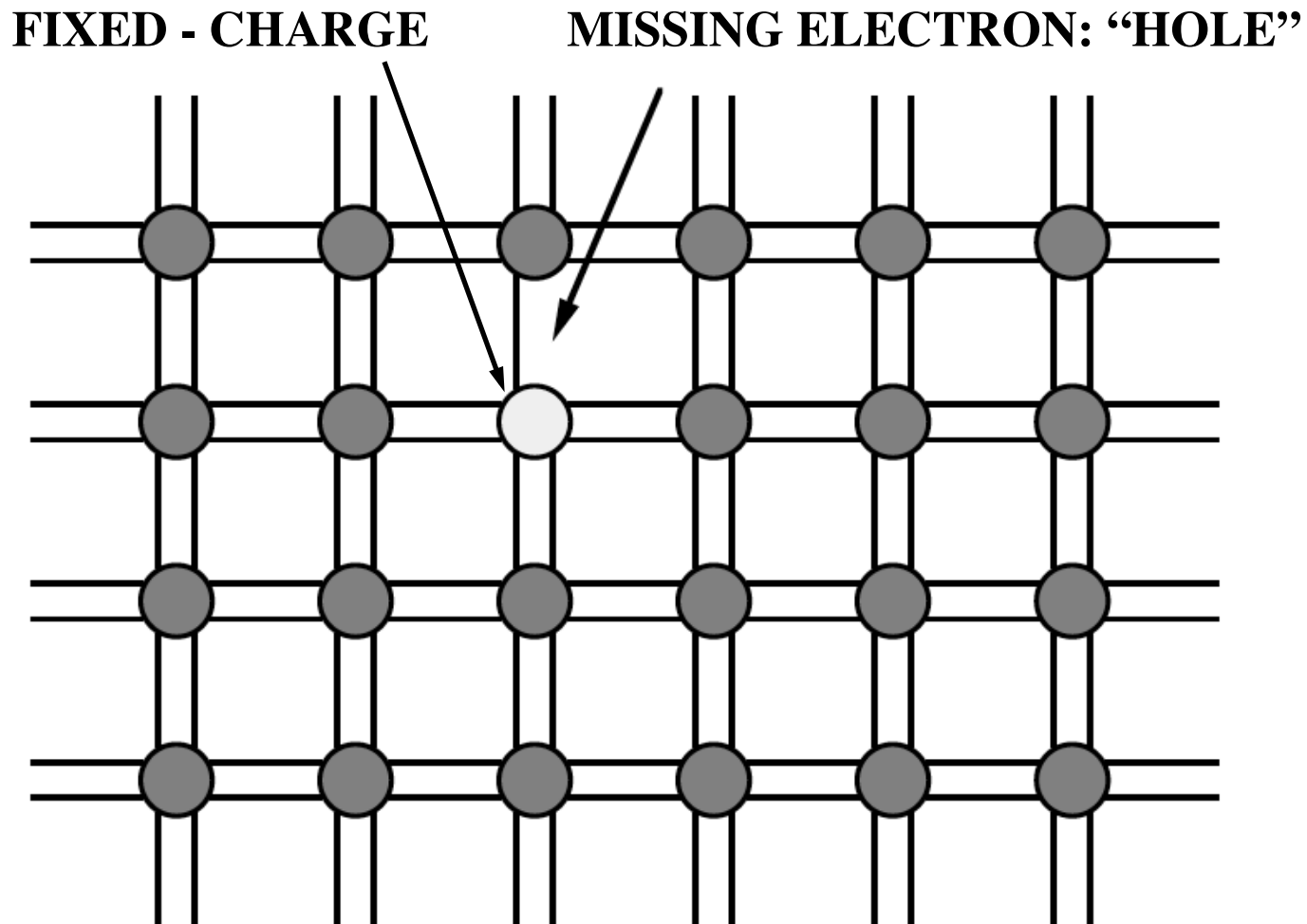
Donor: Phosphorous

- **Donates extra electron: mobile**
- **Also extra proton: Fixed +positive charge**
- **More mobile negative charges: n-type**



Acceptor: Boron

- **Vacancy (“hole”)** that can accept an electron
- **Also missing proton: Fixed negative charge**
- **More mobile positive charges: p-type**



Caution

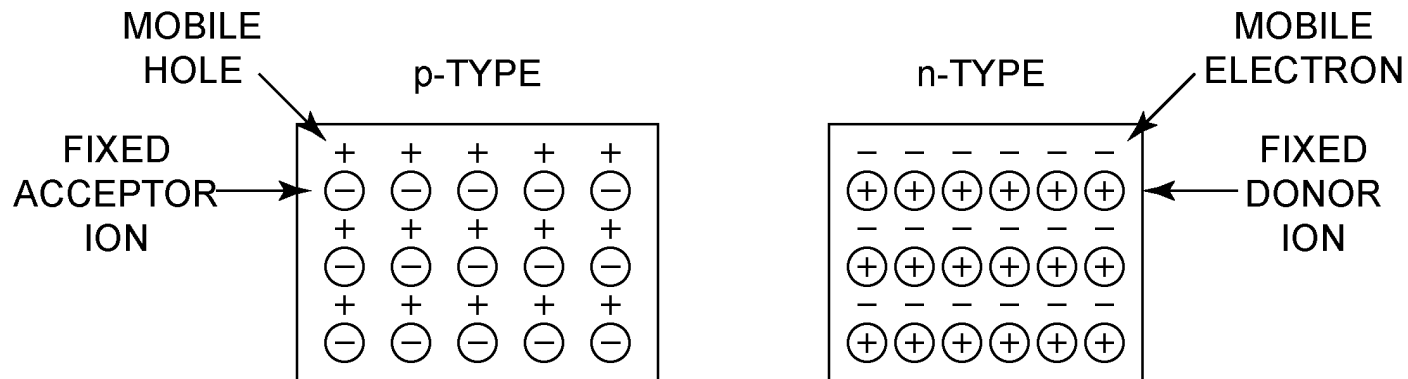
- **Entire semiconductor is electrically neutral**
- **Donor: extra proton in nucleus**
- **Acceptor: missing proton in nucleus**
(for both, relative to Si)

P-N Junction Diode

- **Junction between p-type, n-type materials**
- **Assume structure is simple enough that we can consider only one spatial dimension**
- **Model: only consider “extra” charges**
- **Mobile charges contributed by dopants: each mobile charge was contributed by a fixed charge of opposite sign**

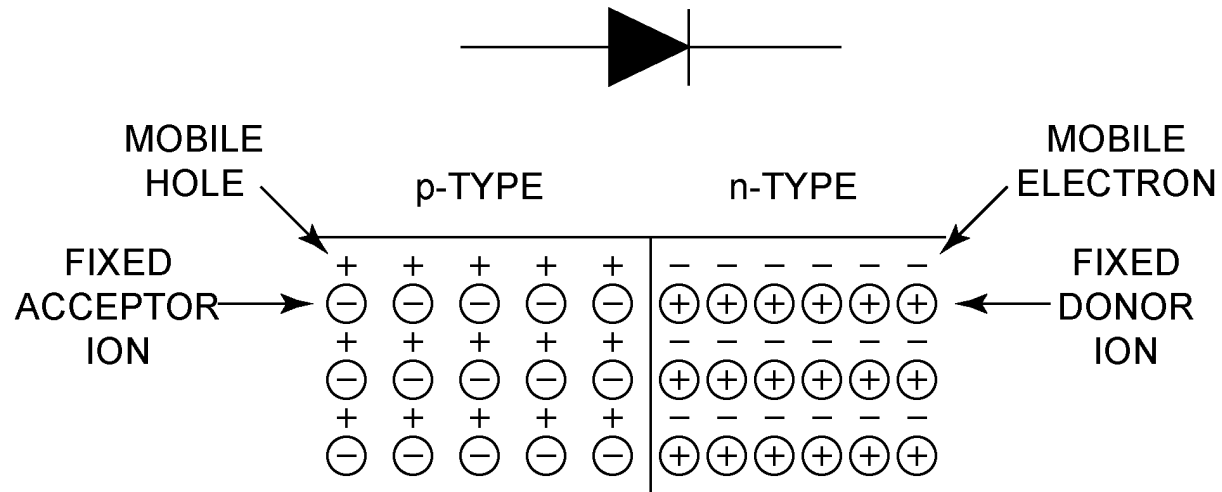
P-N Junction Diode

- **Thought experiment: take isolated p-type, n-type and bring together**



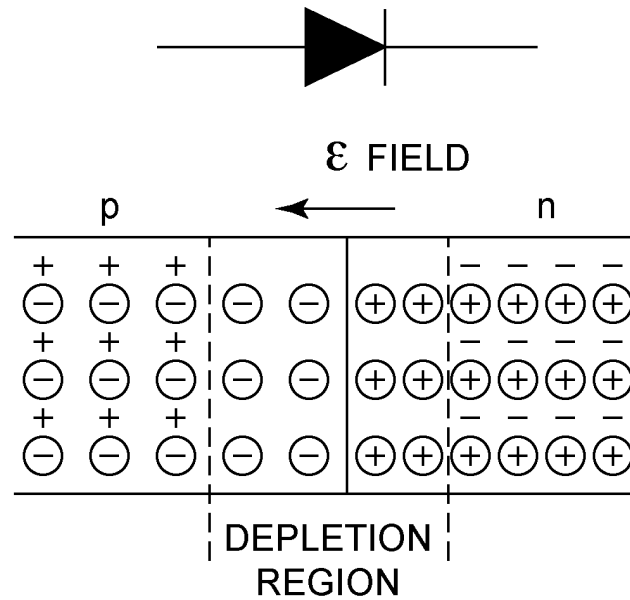
$$V_A = 0 \text{ (t=0)}$$

- **No electric field; Thermal diffusion**



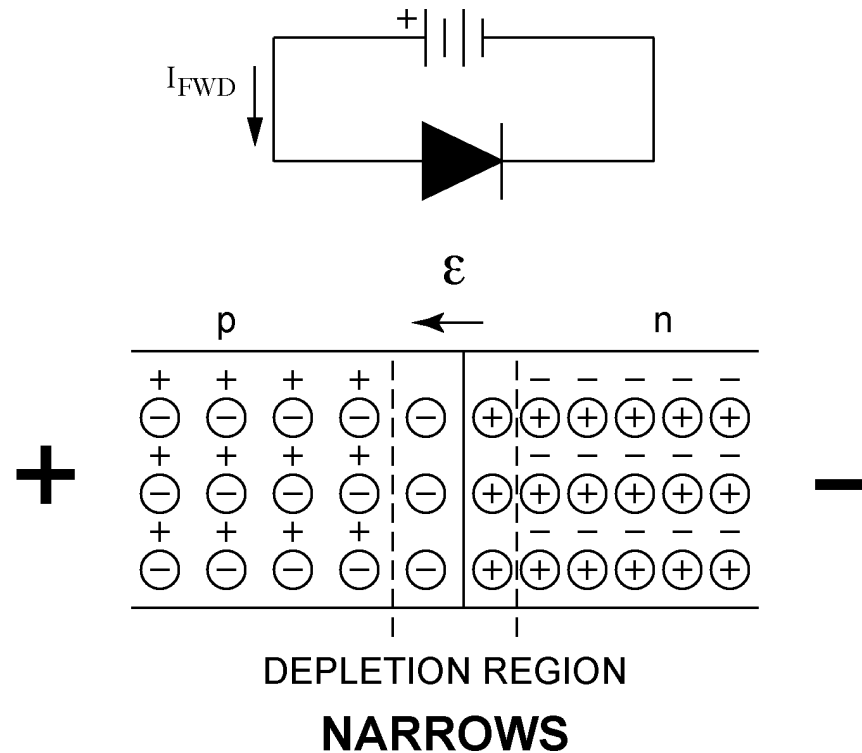
$$V_A = 0 \text{ (equilibrium } t > 0)$$

- **Internal electric field balances diffusion**



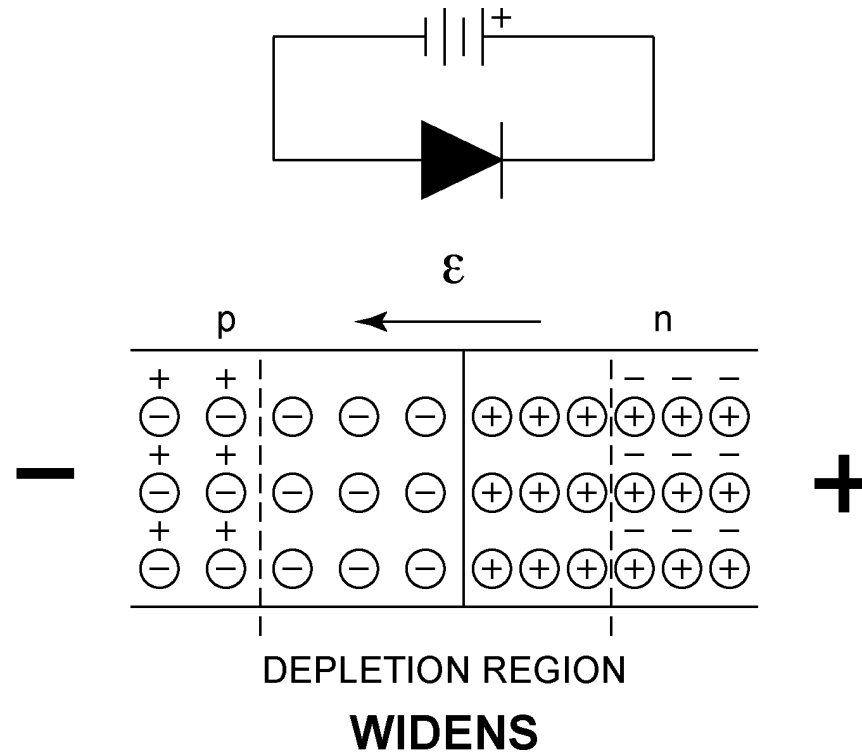
V_A positive (forward bias)

- **Applied V_A “overpowers” internal E field**



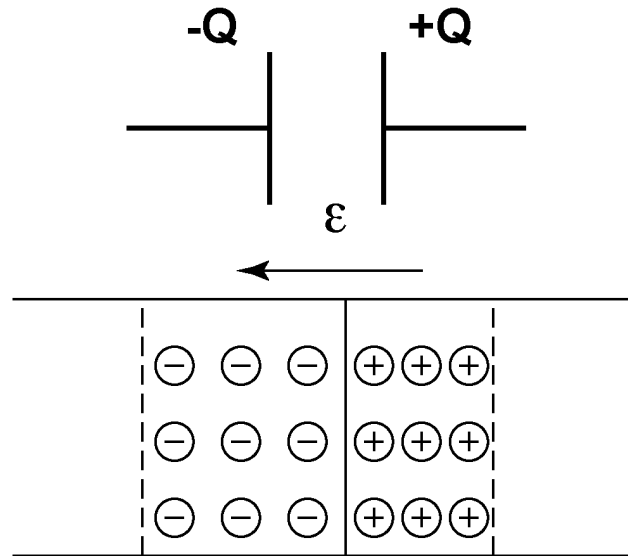
V_A negative (reverse bias)

- **Applied V_A “reinforces” internal E field**
- **Field “pulls” mobile charges further apart**



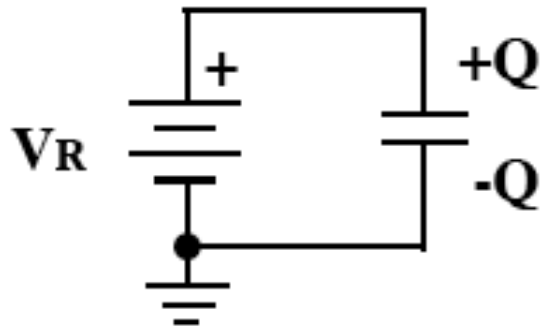
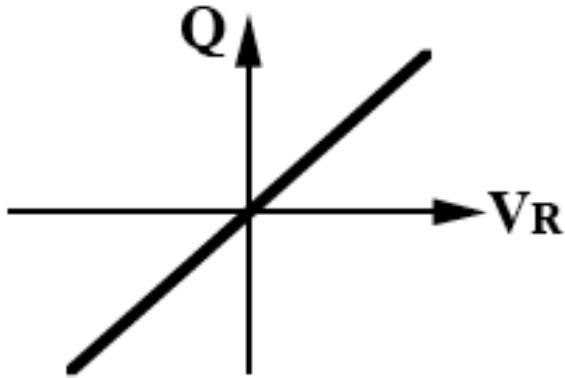
Junction Capacitance

- **Equal and opposite charges, physically separated, changes with voltage: Capacitance!**



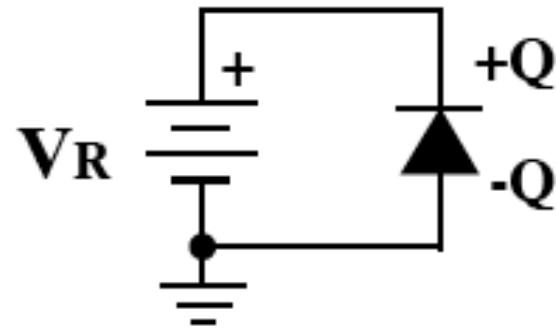
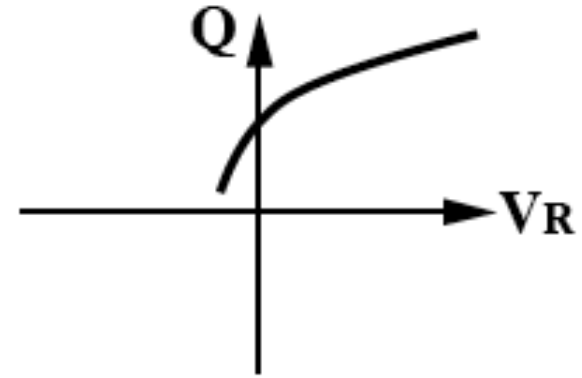
Junction Capacitance

$$C = dQ/dV$$



Same slope for any V

Linear capacitance



Slope depends on V

Nonlinear capacitance

Semiconductor Physics Review Summary

- **Conduction in semiconductor:
e- and "holes" (absence of electrons)**
- **N region has mostly mobile electrons;
P region has mostly mobile "holes"**
- **PN junction has a depletion region
(no mobile carriers)**
- **Size of the depletion region depends on the
applied voltage and the P, N doping**
- **Reverse biased PN junction acts like a
nonlinear capacitor**